Safety Assessment of Polyamino Sugar Condensate as Used in Cosmetics

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The Expert Panel for Cosmetic Ingredient Safety members are: Chair, Wilma F. Bergfeld, M.D., F.A.C.P.; Donald V. Belsito, M.D.; David E. Cohen, M.D.; Curtis D. Klaassen, Ph.D.; Allan E. Rettie, Ph.D.; David Ross, Ph.D.; Thomas J. Slaga, Ph.D.; Paul W. Snyder, D.V.M., Ph.D.; and Susan C. Tilton, Ph.D. The Cosmetic Ingredient Review (CIR) Executive Director is Bart Heldreth, Ph.D. This safety assessment was prepared by Preethi Raj, Senior Scientific Analyst/Writer, CIR.

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Memorandum

To:	CIR Expert Panel Members and Liaisons
From:	Preethi S. Raj, M.Sc.,
	Senior Scientific Analyst/Writer, CIR
Date:	February 10, 2023
Subject:	Re-Review of the Safety Assessment of Polyamino Sugar Condensate

The Expert Panel for Cosmetic Ingredient Safety (Panel) first published a review of the safety of Polyamino Sugar Condensate in 1982 (identified in the pdf as *originalreport_PolyaminoSugarCondensate_032023*). On the basis of the available animal data and limited human experience presented in the report, the Panel concluded that in the present practices of use and concentration (described in the safety assessment), Polyamino Sugar Condensate is safe for topical application to humans. The Panel previously considered a re-review of this report and reaffirmed the 1982 conclusion, as published in 2005 (*rereview2005_PolyaminoSugarCondensate_032023*).

Because it has been at least 15 years since the previous re-review was published, in accordance with Cosmetic Ingredient Review (CIR) Procedures, the Panel should consider whether the safety assessment of Polyamino Sugar Condensate should be re-opened. An extensive search of the world's literature was performed for studies dated 2000 forward. No relevant published data were found. An historical overview, comparison of original and new use data, and the search strategy used are included herein (*newdata_PolyaminoSugarCondensate_032023*).

Also included for your review is a table of current and historical use data (*usetable_PolyaminoSugarCondensate_032023*). (As per the Panel's request at the December 2022 meeting, an updated use table format has been implemented. The frequency and concentration of use is presented both cumulatively by likely duration and exposure and individually by product category.) According to 2022 FDA VCRP data, Polyamino Sugar Condensate has 1 reported use in a body and hand formulation. At the time this ingredient was last considered for re-review, 25 uses were reported. Concentration of use data were neither reported during the last review nor in response to a survey conducted by the Council in 2022.

If upon review of the updated use data the Panel determines that a re-review is warranted, a Draft Amended Report will be presented at an upcoming meeting.

Re-Review - Polyamino Sugar Condensate - History and New Data

(Preethi Raj – March 2023 meeting)

Ingredients (1)	Citation	Conclusion	Use - New Data	Results	Use - Existing Data	Results	Notes
Polyamino Sugar	JACT 1(4):25-32, 1982	safe for topical	frequency of use (2022)	1 use	frequency of use (2001)	25 uses	-frequency of use decreased
Condensate		application	conc of use (2022)	NR	conc of use (2001)	NR	-concentrations of use were not reported in 2001
<u>Changes to Original List</u> none	IJT 24(S1):80-81, 2005	reaffirmed					or in 2022

NOTABLE NEW DATA						
Publication Study Type Results – Brief Overview Different from Existing Data?						
no new published data						

Search (from 2000 on)

Pubmed

(((polyamino sugar condensate) OR (120022-92-6)) OR (aqualizer ej)) OR (aminoic acid sugar mixture) AND (2000:2023[pdat]) – 2,799 hits/0 useful

General Search

polyamino sugar condensate toxicity - 165,000 hits/0 useful

Fable 1. 2022 and historical free	equency and concentration of	of use according to likely	y duration and ex	posure and produc	t category

	# of Uses		Max Conc	of Use (%)				
	2022 ¹	2001 ²	2022 ³	2001 ²				
Totals	1	25	NR	NR				
summarized by likely duration and exposure*								
Duration of Use								
Leave-On	1	22	NR	NR				
Rinse-Off	NR	3	NR	NR				
Diluted for (Bath) Use	NR	NR	NR	NR				
Exposure Type**								
Eye Area	NR	4	NR	NR				
Incidental Ingestion	NR	NR	NR	NR				
Incidental Inhalation-Spray	1ª	5 ^a ; 11 ^b	NR	NR				
Incidental Inhalation-Powder	1ª	5ª	NR	NR				
Dermal Contact	1	25	NR	NR				
Deodorant (underarm)	NR	NR	NR	NR				
Hair - Non-Coloring	NR	NR	NR	NR				
Hair-Coloring	NR	NR	NR	NR				
Nail	NR	NR	NR	NR				
Mucous Membrane	NR	NR	NR	NR				
Baby Products	NR	NR	NR	NR				
as reported by product category								
Eye Makeup Preparations								
Other Eye Makeup Preparations	NR	4	NR	NR				
Skin Care Preparations								
Cleansing	NR	2	NR	NR				
Face and Neck (exc shave)	NR	1	NR	NR				
Body and Hand (exc shave)	1	4	NR	NR				
Moisturizing	NR	9	NR	NR				
Night	NR	1	NR	NR				
Paste Masks (mud packs)	NR	1	NR	NR				
Other Skin Care Preparations	NR	2	NR	NR				
Suntan Preparations								
Suntan Gels, Creams, and Liquids	NR	1	NR	NR				

NR - not reported

*likely duration and exposure is derived based on product category (see Use Categorization https://www.cir-safety.org/cir-findings)

**Because each ingredient may be used in cosmetics with multiple exposure types, the sum of all exposure types may not equal the sum of total uses.

^a Not specified whether a spray or a powder, but it is possible the use can be as a spray or a powder, therefore the information is captured in both categories

^b It is possible these products are sprays, but it is not specified whether the reported uses are sprays.

REFERENCES

- U.S. Food and Drug Administration Center for Food Safety & Applied Nutrition (CFSAN). 2022. Voluntary Cosmetic Registration Program - Frequency of Use of Cosmetic Ingredients (VCRP). Obtained under the Freedom of Information Act from CFSAN; requested as "Frequency of Use Data" January 4, 2022; received January 11, 2022.
- Cosmetic Ingredient Review Expert Panel. Annual Review of Cosmetic Ingredient Safety Assessments--2002/2003. Int J Toxicol. 2005;24 Suppl 1:1-102.
- Personal Care Products Council. 2022. Concentration of Use by FDA Product Category: Polyamino Sugar Condensate. Unpublished data submitted by the Personal Care Products Council on October 24, 2022.

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JOURNAL OF THE AMERICAN COLLEGE OF TOXICOLOGY Volume 1, Number 4, 1982 Mary Ann Liebert, Inc., Publishers

Final Report of the Safety Assessment of Polyamino Sugar Condensate

Polyamino Sugar Condensate (PSC) is the product of a condensation reaction between amino acids and sugars. It appears in over 100 cosmetic preparations at concentrations up to 1%.

PSC has an acute oral toxicity greater than 5 g/kg in rats. In tests on rabbits, undiluted PSC was not a primary irritant and produced only mild irritation in some animals. Subacute skin irritation was not observed in rabbits when PSC (undiluted) was applied. Human safety data indicate that PSC is nonsensitizing and, at worst, a mild irritant. PSC is also nonphototoxic.

On the basis of the available animal data and limited human experience, it is concluded that Polyamino Sugar Condensate is safe for topical application to humans.

CHEMICAL AND PHYSICAL PROPERTIES

Structure

POLYAMINO Sugar Condensate (PSC) is a sugar-amino acid condensation product which conforms to the following structure:⁽¹⁾



where R is an amino acid alkyl group and R' is a monosaccharide ring.

Three patients describe the extraction and/or preparation of the condensate.⁽²⁻⁴⁾ It is prepared commercially by reacting water soluble salts of amino acids with ribose, fructose, and glucose to form N-glycosides at temperatures limited to 63°C. The mixture of amino acids includes alanine, glycine, leucine, proline, serine, threonine, tyrosine, valine, aspartic acid, glutamic acid, arginine hydrochloride, histidine hydrochloride, lysine hydrochloride, and pyroglutamic acid. A typical condensation reaction would proceed as follows:

where R and R' are as defined above.

Lactic acid is added to form a condensation mixture. Free amino acids, urea, potassium chloride, calcium chloride, and sodium chloride are then added to form the final product.⁽⁵⁾

Polyamino Sugar Condensate was analyzed for sugar, nitrogen and water content. The results are summarized in Table 1.⁽⁵⁾

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COSMETIC INGREDIENT REVIEW

Substance analyzed for	Content in PSC
Sugar	8-13%
Nitrogen	7.5-10.5 μmol/mg
Alpha-amino nitrogen	3.0-6.0 μmol/mg
Water (and volatiles)	10-20%

From Ref. 5.

Physical Properties

PSC is a water soluble, brown semisolid with a pH range of 3.9 to 4.3^(6.7) No other data describing the compound's physical characteristics have been reported.

Impurities

Impurities in a random lot analysis of PSC were: heavy metal (as Pb), 3.5 ppm; arsenic, 0.01 ppm and mercury, 0.0003 ppm.⁽⁶⁾ When a method with a detection limit of 0.1 ppm was used, no nitrite impurities were found in PSC;⁽⁸⁾ nitrosodiethanolamine was not present (detection limit = 0.028 ppm) in either a one-year-old or three-year-old sample of PSC.⁽⁹⁾ With regard to trace amounts of pyrazines and/or volatile N-nitrosamines in PSC, no analytical data verifying their presence or absence were available.

USE

Purpose in Cosmetics

PSC is a moisturizing agent in cosmetic products, especially makeup and skin care formulations.⁽¹⁾

Scope and Extent of Use in Cosmetics

According to industry's voluntary submissions to the Food and Drug Administration in 1976, PSC was used in over 100 formulations up to concentrations of 1% (Table 2). It is estimated that over 40,000,000 units containing this ingredient have been distributed in the last 10 years.⁽⁷⁾

BIOLOGICAL PROPERTIES

General Effects

Discussion of the Maillard reaction is included here because of its possible relationship to the production of PSC. The Maillard reaction, a process which occurs during the browning of foods, involves a complexing of amino acids with reducing sugars to form glycosylamino products, rearrangements of which result in stable Amadori compounds. Nutritional and toxicological studies focus on the problems associated with such compounds.⁽¹¹⁻¹⁵⁾

Studies have shown products of the Maillard reaction have a particular tendency to nitrosate in the presence of sodium nitrite.⁽¹⁶⁻¹⁸⁾ Heyns⁽¹⁹⁾ found that nitrosopiperdine could be formed when D-glucose and D-lysine react at 105°C in the presence of sodium nitrite. While Devik⁽¹⁶⁾ suggests that dimethylnitrosamine may form upon heating various amino acid-glucose combinations to 104°-105°C, two subsequent studies failed to confirm this.^(17,18) Shinohara et al.⁽²⁰⁾ reported that

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Ingredient	Cosmetic product type	Concentration (percent)	Number of product formulations
Polyamino sugar	Other eye makeup preparations	≤0.1	1
condensate	Shampoos (noncoloring)	≤0.1	1
	Blushers (all types)	≤0.1	6
	Face powders	≤0.1	2
	Foundations	>0.1-1	3
		≤0.1	5
	Leg and Body paints	>0.1-1	1
	Lipstick	>0.1-1	1
	Makeup bases	>0.1-1	2
	-	≤0.1	3
	Rouges	≤0.1	4
	Makeup Fixatives	≤0.1	1
	Other makeup preparations	≤0.1	1
	Cuticle softeners	>0.1-1	1
	Nail creams and lotions	≤0.1	1
	After shave lotions	>0.1-1	1
		≤0.1	
	Cleansing (cold creams, cleansing lotions, liquids, and pads)	>0.1-1	6
	Face, body, and hand (excluding	>0.1-1	11
	shaving preparations)	≤0.1	1
	Moisturizing	> 0.1-1	19
	Ū.	≤0.1	7
	Night	>0.1-1	6
	-	≤0.1	14
	Paste masks (mud packs)	>0.1-1	1
		≤0.1	2
	Skin fresheners	>0.1-1	1
	Other skin care preparations	>0.1-1	5
		≤0.1	3
			Total 111

TABLE 2. PRODUCT FORMULATION DATA.^a

^aFrom Ref. 10.

heating a mixture of glucose and lysine at 100°C for 10 h resulted in a browning product which was mutagenic to S. typhimurium TA 100.

Using the Ames test, nitrosated Amadori compounds were determined to be mutagenic in the presence of sodium nitrite or under prolonged heating.^(21,22)

In general, the Maillard reaction is favored by specific reaction conditions; these include high temperatures (greater than 100°C), lack of moisture, and high pH, which in general are not expected to occur during the manufacture of PSC.⁽¹¹⁾

In the preparation of PSC, the reaction is kept at or below 63° C and in aqueous solution at all times. Lactic acid is added in the final stages of PSC preparation, so that the pH of the product is adjusted between 3.8 and 4.3.⁽⁷⁾

The formation of nitroso compounds from Maillard products requires either heating to a temperature higher than 100°C or those products being in the presence of nitrosating agents. The preparation of PSC is limited to temperatures below 63°C, and analyses indicate that even in material stored up to three years, nitrite (<0.1 ppm) and N-nitrosodiethanolamine (<0.02 ppm) are absent.^(8.9)

COSMETIC INGREDIENT REVIEW

Absorption, Metabolism, and Excretion

When administered intravenously to humans, heat sterilized glucose-amino acid complexes were completely excreted in the urine; the complexes were not resorbed from the glomerular filtrate. When they were administered by gastric tube, these same substances were not detected in the subjects' blood or urine, indicating a lack of absorption.⁽¹³⁾

An amino acid-glucose mixture was browned at 37°C, stored for one month and then fed to an unspecified number of weanling Sprague-Dawley rats for 22 days. As a result of the unavailability of essential and nonessential amino acids as a nitrogen source for protein building, these rats experienced no weight gain. Moreover, during the experiment the animals also experienced decreased utilization of plasma amino acids; not only was the incorporation of free amino acids into proteins affected, but also the catabolic rates of some amino acids may have been decreased.^(23,24)

Animal Toxicology

Acute Oral Toxicity

When administered by gastric intubation, the acute oral LD50 of PSC was determined to be >5 g/kg. This dosage (5 g/kg) was administered to 10 rats, and observations were made at 1, 3, and 6 h and daily for two weeks. During this period, no toxic effects were reported. It was concluded that under these conditions PSC is nontoxic.⁽²⁵⁾

Acute Skin Irritation

Two studies used the Draize method to test the potential irritancy of PSC on rabbit skin. In these tests, 0.5 g undiluted PSC was applied under occlusive patches both to intact and to abraded skin of six rabbits. Patches remained in place for 24 h, during which time the animals were immobilized. Evidence of irritation was scored at 24 and 72 h. In neither study were there any signs of irritation to either intact or abraded skin (Primary Irritation Indices = 0). Consequently, PSC was not considered to be a primary skin irritant.^(26,27)

Eye Irritation

Three studies, also using the Draize method, reported on the potential irritancy of undiluted PSC to the eyes of rabbits.

One of the six rabbits tested in the first study exhibited conjunctival redness at 24 h, resulting in a mean irritation score of 0.33 (where the maximum possible score is 110). All eyes appeared normal at 48 and 72 h. Thus this study concluded that undiluted PSC caused "insignificant irritation."⁽²⁸⁾

In the second study, conjunctival redness was observed in the same four of six rabbits at 24, 48, and 72 h. Mean scores for these observation periods were 3.33, 1.67, and 1.67, respectively (where the maximum possible score is 110). At seven days no irritation was observed in any animal. In this test, PSC was determined to be mildly irritating.⁽²⁹⁾

In the final study, the eyes of half of the six animals tested were washed with 20 ml warm water four seconds after instillation of PSC. At no time was any irritation observed in either washed or unwashed eyes. PSC was considered to be nonirritating under the test conditions.⁽³⁰⁾

Subchronic Skin Irritation

PSC was tested in eight rabbits for subchronic skin irritation. Animals were depilated on the back and flanks, and half the skin site abraded in each rabbit. Four dose levels were assigned, with two animals per dose; doses were applied daily for 20 days. The animals were checked daily for changes in body weight, behavior, and food consumption, as well as for signs of skin irritation. Likewise, urine and blood were analyzed for abnormalities. Undiluted PSC was applied to each group of rabbits in doses of 0, 0.5, 1.0, or 1.5 g/kg. No irritation developed in either intact or abraded skin, and no physiological abnormalities were observed. It was concluded that undiluted PSC caused no systemic toxicity by percutaneous absorption.⁽³¹⁾

Sensitization

A gel makeup containing 0.10% PSC and a cleansing cream containing 0.12% PSC were tested on groups of 10 and six guinea pigs, respectively, for potential sensitization. Guinea pigs were

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depilated, and 0.1 ml of a 1% suspension (in saline) was injected intracutaneously. Injections were made every other day, until there was a total of 10 injections per animal. Two weeks after the last sensitizing injection, 0.05 ml of the suspension was injected in a fresh skin site as a challenge. Observations were made 24 h after each injection and a comparison made between challenge and sensitizing reactions. Each study showed no appreciable differences between the two types of reactions. The cosmetic products containing 0.10 and 0.12% PSC were found to be nonsensitizing.^(32,33)

Clinical Assessment of Safety

Skin Irritation and Sensitization

Single insult patch test

A single insult patch test was performed on 55 subjects; 0.5 ml of undiluted PSC was applied to their skin. Observations made at 24 and 48 h revealed no irritation in any subject caused by the application of PSC (though one case of adhesive tape dermatitis did arise).⁽³⁴⁾

Repeat insult patch test

Four studies tested PSC and products containing PSC for potential irritation and sensitization on human skin. In these tests, various concentrations of PSC were placed under occlusive patches, left in place for 24 h, and then removed. Twenty-four hours later, sites were scored and a new patch was applied; this procedure was repeated until 10 induction patches had been applied. Ten to 14 days after the final induction patch had been removed, a challenge patch was placed on the original test site and/or to a fresh site adjacent to that. Sites were scored 24 and 48 h after these challenge patch applications. Results of these tests appear in Table 3.^(35,36)

A moisturizing lotion containing 0.24% PSC and a solution of 10% PSC were tested on 201 and 51 subjects, respectively. Concentrations of up to 10% PSC did not result in any reactions.^(35,36)

A solution containing 30% PSC caused a single reaction in two of 54 subjects tested. One subject experienced slight erythema after the third exposure, while erythema and edema were observed in another subject after the second exposure. These people did not have any reactions to the other patches.⁽³⁶⁾

On 10 occasions a hand lotion containing 3.0% PSC was applied to 51 subjects. Four subjects experienced irritation which ranged from slight erythema (1 + reaction) to moderate erythema, edema, and vesicle formation (3 + reaction). Irritation occurred after a single exposure and then disappeared, except in one case when a 1 + (erythema) reaction was observed after exposure to patches 2, 6, 7, and 9.⁽³⁶⁾

Overall, the skin irritation and sensitization tests indicated that a 10% PSC solution (aqueous) and a cosmetic lotion containing 0.24% PSC were nonirritating, while a 30% PSC solution (aqueous) and a cosmetic lotion containing 3% PSC were mildly irritating. In none of the studies were there any reactions to challenge patches; thus, these tests determined that PSC was nonsensitizing at test concentrations.^(35,36)

			M/F	Age range	No. of I	<u> </u>	
Product/ Ingredient	PSC Conc.	No. of subjects			Patches 1-10 (Challenge Patch	Comments
Moisturizing							
lotion	0.24%	201	57/144	8-73	"No untow	ard effects"	Nonirritant
PSC	10.0% (aq)	51	11/40	12-70	0	0	Nonirritant
PSC	30.0% (ag)	54	12/42	12-70	2	0	1+,2+
Hand lotion	3.0%	51	11/40	12-70	4	0	1+-3+

TABLE 3. REPEAT INSULT PATCH TESTS USING PSC.^a

^aData from Refs. 35 and 36.

COSMETIC INGREDIENT REVIEW

Phototoxicity

A 30% (aqueous) solution of PSC (0.2 ml) was applied to 10 subjects. Patches were left in place for 24 h and then removed. One site was used as a control, while the other was irradiated at 4400 μ W/cm² through the use of four f4BL black-light tubes having a peak output of 360 nm. Observations made at 24, 48, and 96 h and at seven days revealed one subject with mild erythema at both control and irradiated sites. PSC (30%) did not produce phototoxicity in the subjects who were studied.⁽³⁷⁾

SUMMARY

Polyamino Sugar Condensate (PSC) is the product of a condensation reaction between amino acids and sugars. It appears in over 100 cosmetic preparations at concentrations up to 1%.

Many nutritional and toxicological studies have dealt with the Maillard reaction, a process that occurs when foods are browned. This reaction appears to be unrelated to the production of PSC; however, specific analytical data are unavailable to confirm this statement. Data regarding the absence in PSC of traces of mutagenic agents and volatile N-nitrosamines are also unavailable.

PSC has an acute oral toxicity greater than 5 g/kg in rats. When it was applied undiluted to rabbit skin, it was found not to be a primary irritant. Application of PSC to the eyes of rabbits produced mild irritation in some animals; this usually appeared as conjunctival redness. Subacute skin irritation was not observed in rabbits when PSC (undiluted) was applied.

No data were available on the teratogenicity, mutagenicity or carcinogenicity of PSC.

In a single-insult patch test, undiluted PSC applied to 55 subjects did not cause irritation. In a number of repeated insult patch tests, an aqueous solution containing 10% PSC and a moisturizer containing 0.24% PSC caused no irritation in 51 and 201 subjects, respectively. To test an aqueous solution of 30% PSC, a total of 594 patches were applied to 54 subjects; two subjects experienced single reactions. A hand lotion containing 3% PSC resulted in four reactions in the 51 tested subjects, to whom a total of 561 patches were applied. In all tests, there were no reactions to challenge patches. Thus, human safety data indicate that PSC is nonsensitizing and, at worst, a mild irritant. PSC is also nonphototoxic.

CONCLUSION

On the basis of the available animal data and limited human experience presented in this report, the Panel concludes that in the present practices of use and concentration, Polyamino Sugar Condensate is safe for topical application to humans.

ACKNOWLEDGMENT

Mr. Kevin Fisher, Scientific Analyst and writer prepared the Technical Analysis used by the Expert Panel in developing this report.

REFERENCES

- 1. CTFA. (July 1, 1979). Submission of data by CTFA in support of safety of Polyamino Sugar Condenste.*
- 2. JACOBI, K.O. (May 8, 1962). Process for removing the water soluble materials from a keratin structure and cosmetic or pharmaceutical product formed therefrom. U.S. Patent Office, No. 3,033,755.
- 3. JACOBI, K.O. (July 13, 1965). Process for removing and purifying water soluble constituents from a keratin structure. U.S. Patent Office, No. 3,194,737.
- 4. JACOBI, K.O. (Jan. 25, 1966). Method of increasing water-absorbing ability of human skin and composition therefrom. U.S. Patent Office, No. 3,231,472.

^{*}Available upon request: Administrator, Cosmetic Ingredient Review, Suite 810, 1110 Vermont Ave., NW, Washington, DC 20005.

- 5. KOLMAR LABORATORIES. (Feb. 23, 1979). Submission of data by CTFA in support of safety of Polyamino Sugar Condensate.*
- 6. BIOSAFETY LABORATORIES. (Feb. 9, 1979). Submission of data by CTFA in support of safety of Polyamino Sugar Condensate.*
- 7. CTFA. (July 1, 1979). Submission of data by CTFA in support of safety of Polyamino Sugar Condensate. CTFA Chemical Description.*
- 8. THERMOELECTRON CORP. (June 29, 1979). Submission of data by CTFA in support of safety of Polyamino Sugar Condensate.*
- 9. THERMOELECTRON CORP. (July 1, 1979). Submission of data by CTFA in support of safety of Polyamino Sugar Condensate.*
- 10. FDA. (Aug. 31, 1976). Cosmetic product formulation data. Washington, DC: Food and Drug Administration.
- 11. ADRIAN, J. (1974). Nutritional and physiological consequences of the Maillard Reaction. World Rev. Nutr. Diet. 11, 71-122.
- 12. FREEMAN, J.B., STEGINK, L.D., MEYER, P.D., FRY, L.K., and DENBESTEN, L. (1975). Excessive urinary zinc losses during parenteral alimentation. J. Surg. Res. 18, 463.
- 13. STEGINK, L.D. and PITKIN, R.M. (1977). Placental transfer of glucose-amino acid complexes present in parenteral solutions. Am. J. Clin. Nutr. 30(7), 1087-93.
- 14. FINK, H. (1963). Die experiementelle alimentar Lebernekrose, als empfindlicher Indikator bein thermischer Belastung der Milch uber Magermilchtrocknung. Nahrung 7, 277–99.
- 15. FINK, H., SCHLIE, I., and RUGE, U. (1958). Uber ernahrungsphysiologische Veranderungen der Milch beim technischen Trocknen. Z. Naturforsch 13b, 610-16.
- 16. DEVIK, O.G. (1967). Formation of N-nitrosamines by the Maillard reaction. Acta. Chem. Scand. 21(8), 2302-3.
- 17. HEYNS, K. and KOCH, H. (1970). Zur Frage der Entstehung von Nitrosaminen bei der Reacktion von Monosacchariden mit Aminosauren (Maillard-Reaktion). Tetrahedron Lett. 10, 741-4.
- 18. HEYNS, K. and ROPER, H. (1974). Gas chromatographic trace analysis of volatile nitrosamines in various types of wheat flour after application of different nitrogen fertilisers to the wheat. IARC Int. Agency Res. Cancer Sci. Publ. 9, 166-72.
- HEYNS, K. and ROPER, H. (1974). Zur Frage der Entstehung von Nitrosaminen bein der Reaktion von Monosacchariden mit Aminosauren (Maillard-Reaktion) in Gegenwart von Natriumnitrit. 2. Mitteilung. Z. Lebensmitt. Unter.-Such. 154(4), 193-200.
- 20. SHINOHARA, K., WU, PIT., JAHAN, N., TANAKA, M., MORINAGA, W., MURAKAMI, H., OMURA, H. (1980). Mutagenicity of the browning mixtures by amin-carbonyl reactions on Salmonella typhimurium TA 100. Agr. Biol. Chem. 44(3), 671-672.
- 21. IWAKOKA, W.T. and MEAKER, E.H. (Apr. 1-6, 1979). Formation of mutagens in the cooking of commercially prepared foods (Abstr.). ACS/CSJ Chemical Congress, Honolulu, Hawaii.
- 22. COUGHLIN, J.R. (1979). Formation of N-nitrosamines from Maillard browning reaction products in the presence of nitrite. Diss. Abstr. Int. [B], 40(2), 719.
- 23. SGARBIERI, V.C., AMAYA, J., TANAKA, M., and CHICHESTER, C.O. (1973). Physiological consequences of feeding to rat a browned synthetic amino acid-sugar mixture (Maillard reaction). Arch. Latinoam. Nutr. 23(3), 363-78.
- 24. SGARBIERI, V.C., AMAYA, J., TANAKA, M., and CHICHESTER, C.O. (1973). Nutritional consequences of the Maillard reaction. Amino acid availability from the fructose-leucine and fructosetryptophan in the rat. J. Nutr. 103 (5), 657-63.
- 25. BIOSAFETY LABORATORIES. (Feb. 26, 1979). Submission of data by CTFA in support of safety of Polyamino Sugar Condensate. Acute oral toxicity.*
- 26. KOLMAR LABORATORIES. (1960). Submission of data by CTFA in support of safety of Polyamino Sugar Condensate. Primary skin irritation.*
- 27. BIOSAFETY LABORATORIES. (Feb. 26, 1978). Submission of data by CTFA in support of safety of Polyamino Sugar Condensate. Primary skin irritation.*
- 28. BIOSAFETY LABORATORIES. (Feb. 26, 1979). Submission of data of CTFA in support of safety of Polyamino Sugar Condensate. Acute eye irritation.*
- 29. FDA RESEARCH LABORATORIES: (Feb. 16, 1979). Submission of data by CTFA in support of safety of Polyamino Sugar Condensate. Acute eye irritation.*
- 30. KOLMAR LABORATORIES. (Oct. 1960). Submission of data by CTFA in support of safety of Polyamino Sugar Condensate. Acute eye irritation.*

- 31. KOLMAR LABORATORIES: (Sept. 8, 1965). Submission of data by CTFA in support of safety of Polyamino Sugar Condensate. Subacute skin irritation.*
- 32. KOLMAR LABORATORIES. (Sept. 8, 1965). Submission of data by CTFA in support of safety of Polyamino Sugar Condensate. Sensitization.*
- 33. KOLMAR LABORATORIES. (Dec. 9, 1969). Submission of data by CTFA in support of safety of Polyamino Sugar Condensate. Sensitization.*
- 34. KOLMAR LABORATORIES. (Oct. 1960). Submission of data by CTFA in support of safety of Polyamino Sugar Condensate. Clinical assessment/single insult.*
- 35. CTFA. (June 29, 1979). Submission of data by CTFA in support of safety of Polyamino Sugar Condensate. Clinical assessment/repeated insult.*
- 36. FDA RESEARCH LABORATORIES. (March 14, 1979). Submission of data by CTFA in support of safety of Polyamino Sugar Condensate. Clinical assessment/repeated insult.*
- 37. FDA RESEARCH LABORATORIES. (March 16, 1979). Submission of data by CTFA in support of safety of Polyamino Sugar Condensate. Clinical assessment/phototoxicity.*

- Cosmetic, Toiletry, and Fragrance Association (CTFA). 2002. Use concentration data on peg stearates from industry survey. Unpublished data submitted by CTFA, September 13, 2002. 5 pages.¹⁹
- Cutler, R. R., P. Wilson, and F. V. Clarke. 1987. The effect of polyoxyethylene stearate (POES) on the growth of mycobacteria in radiometric 7H12 Middlebrook TB medium. *Tubercle*. 68:209–220.
- Dahlgren, C., and K. E. Magnusson. 1980. Modulation of polymorphonuclear leukocyte locomotion by synthetic amphiphiles. *Scand. J. Infect. Dis. Suppl.* 24:44–47.
- Dahlgren, C., I. Rundqvist, O. Stendahl, and K. E. Magnusson. 1980. Modulation of polymorphonuclear leukocyte locomotion by synthetic amphiphiles: Effect of saturated fatty acid esters (C2-C18) of poly(ethyleneglycol) 6000. *Cell Biophys.* 2:253–267.
- Elder, R. L., ed. 1983. Final report on the safety assessment of PEG-2, -6, -8, -12, -20, -32, -40, -50, -100, and -150 Stearates. J. Am. Col. Toxicol. 2:17–34.
- Gowan, W. G., F. Tio, K. Lamp, C. Gowan, and S. Stavchansky. 1986. The effects of Tween 80 MYRJ-52 and polyethylene glycol 8000 on the rate of pentobarbital disappearance cumulative water flux and histological changes in the rat small intestine. *Int. J. Pharm.* 29:169–176.
- Hachiya, N. 1987. Evaluation of chemical genotoxicity by a series of short-term tests. *Akita. Igaku.* 14:269–292.
- Ismail, S., A. A. Mohamed, and M. G. Abd El-Mohsen. 1989. Modifying effects of polyoxyethylene stearates on the rate of acid-catalyzed hydrolysis of acetylsalicyclic acid. *Bull. Pharm. Sci.* 12:52–67.
- Ivanchenko, O. B., O. N. I'inskaya, A. N. Fattakhova, I. M. Skipina, and IE. Cherepneva. 1992. Assessment of genotoxicity of non-ionogenic surfactants. *Biol. Nauki*, 10:75–80.
- Kaur, R., D. J. W. Grant, and T. Eaves. 1980. Comparison of polyethylene glycol and polyoxyethylene stearate as excipients for solid dispersion systems of griseofulvin and tolbutamide. Part 1. Phase equilibria. Part 2. Dissolution and solubility studies. J. Pharm. Sci. 69:1317–1326.
- Kilbanov, A. L., K. Maruyama, V. P Torchilin, and L. Huang. 1990. Ampipathic polyethyleneglycols effectively prolong the circulation time of liposomes. *FEBS Lett.* 268:235–237.
- Kobayashi, I. 1989. The screening test for mutagenesis of surfactants, detergents and fluorescent whitening agents. *Gosei Senzai Kenkyukaishi* 12:19–28.
- Lee, M. H., C. D. Kim, and H. J. Ahn. 1994. Evaluation of surfactant cytotoxicity potential by neutral red uptake assay, MTT assay and cell protein assay. *Korean J. Toxicol.* 10:215–220.
- Lewis, R. W., J. C. McCall, P. A. Botham, and R. Trebilcock. 1994. A comparison of two cytotoxicity tests for predicting the ocular irritancy of surfactants. *Toxicol. In Vitro* 8:867–869.
- Margarit, M. V., I. C. Rodriguez, and A. Cerezo. 1992. Myrj 51 as a suppository excipient in fluence on pharmaceutical availability and bioavailability of sodium valproate. *Int. J. Pharm.* 81:67–73.
- Miller, M. A., L. Thibert, F. Desjardins, S. H. Siddiqi, and A. Dascal. 1996. Growth inhibition of Mycobacterium tuberculosis by polyoxyethylene stearate present in the BACTEC pyrazinamide susceptibility test. J. Clin. Microbiol. 34:84–86.
- Pepe, R. C., J. A. Wenninger, and G. N. McEwen Jr., eds. 2002. International cosmetic ingredient dictionary and handbook, 9th ed., 1201–1207. Washington, DC: CTFA.
- Pienaar, E. W., N. P. Kriek, B. Boneschans, and H. A. Koelman. 1994. Adverse effects of polyethylene glycol and phenytoin on the rectal mucosa of the rat. *Drug Dev. Ind. Pharm.* 20:1493–1502.
- Realini, L., P. Van Der Stufyt, K. De Ridder, B. Hirschel, and F. Portaels. 1997. Inhibitory effects of polyoxyethylene stearate, PANTA, and neutral pH on growth of *Mycobacterium genavense* in BACTEC[®] primary cultures. J. Clin. Microbiol. 35:2791–2794.
- Salminen, S., and E. Salminen. 1987. Urinary excretion of orally administered oxalic acid in MYRJ-45-treated NMRI mice. *Toxicol. Lett.* 37:91–94.

¹⁹Available from the Director, Cosmetic Ingredient Review, 1101 17th Street NW, Suite 310, Washington, DC 20036, USA.

- Satoh, S., Y. Horiguchi, K. Ito, N. Nambu, and T. Nagai. 1985. Irritative effects of suppository bases on the rectal membranes in rabbits. *Arch. Pract. Pharm.* 45:298–303.
- Shen, W.-W., A. G. Danti, and F. N. Bruscato. 1976. Effect of nonionic surfactants on percutaneous absorption of salicylic acid and sodium salicylate in the presence of dimethyl sulfoxide. J. Pharm. Sci. 65:1780–1783.
- Tassett, C., V. Preat, A. Bernard, and M. Roland. 1992. Comparison of nephrotoxicities of different polyoxyethylene formulations of amphotericin B in rats. *Antimicrob. Agent Chemother*. 36:1525–1531.
- Tassett, C., V. Preat, and M. Roland. 1991. The influence of Myrj 59 on the solubility, toxicity and activity of amphotericin B. J. Pharm. Pharmacol. 43:297–302.
- Tatsumi, H., and H. Ritsuko. 1991. A study on the relation between the result of guinea-pig skin irritation test and human patch test with cosmetic ingredients. *Hifu* 33:31–38.
- Uraizee, S. A., A. M. McPhillips, A. A. Sakr, and W. A. Ritschel. 1999. Evaluation of absorption enhancers and enzyme inhibitors for the protection of human insulin: screening study. *Pharm. Ind.* 61:259–262.

POLYAMINO SUGAR CONDENSATE

A safety assessment of Polyamino Sugar Condensate was published in 1982 (Elder 1982). New studies since then (listed at the end of this review), along with the updated information below regarding types and concentrations of use, were considered by the CIR Expert Panel. The Panel determined not to reopen this safety assessment.

In 1976 Polyamino Sugar Condensate was reported to be used in 111 cosmetic preparations with the largest single use occurring in moisturizers at concentrations of $\leq 0.1\%$ to 1%. As reported to the FDA (FDA 2001), Polyamino Sugar Condensate is currently used in 25 cosmetic preparations; however, according to an industry survey (CTFA 2001), Polyamino Sugar Condensate is not currently used in cosmetic preparations. Table 19 presents the available use information.

A study by Peterson et al. (1986) demonstrated that Polyamino Sugar Condensate was absorbed into the stratum corneum and epidermis of athymic nude mice grafted with human skin. Although this information was new, the Expert Panel noted that Polyamino Sugar Condensate is a sugar–amino acid expected to be metabolized to sugars and amino acids. Because the original safety assessment fully discussed the potential production of nitrosamines during the manufacture of Polyamino Sugar Condensate, no further discussion was warranted.

REFERENCES

- CTFA. 2002. Product use concentration information for Polyamino Sugar Condensate; memorandum dated January 22. Unpublished data.²⁰
- Baker, C. G., R. D. Berg, and E. S. Curtis. 1987. Formulating with NMF (normal moisturizing factor). *Drug Cosmet. Ind.* 140:32–39.
- Elder, R. L., ed., 1982. Final report on the safety assessment of Polyamino Sugar Condensate. J. Am. Col. Toxicol. 1:25–32.

²⁰Available from the Director, Cosmetic Ingredient Review, 1101 17th Street NW, Suite 310, Washington, DC 20036, USA.

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Historical and current cosmetic product uses and concentrations for Polyamino Sugar Condensate

Product type	1976 uses (Elder 1982)	2001 uses (FDA 2001)	1976 use concentrations (Elder 1982) (%)	2001 use concentrations (CTFA 2001) (%)
Eye makeup (other)	1	4	<0.1	
Shampoos (noncoloring)	1	_	- <0.1	_
Blushers	6		≤ 0.1	_
Face powders	2		≤ 0.1	_
Foundations	8	_	≤0.1−1	_
Leg and body paints	1		>0.1-1	
Lipstick	1	_	>0.1-1	_
Makeup bases	5	_	$\leq 0.1 - 1$	_
Rouges	4	_	≤0.1	_
Makeup fixatives	1	_	≤0.1	_
Makeup (other)	1	_	≤0.1	_
Cuticle softeners	1	_	>0.1-1	_
Nail creams and lotions	1	_	≤0.1	_
Aftershave lotion	2	_	$\leq 0.1 - 1$	—
Skin-cleansing creams, lotions, liquids, and pads	6	2	>0.1-1	_
Face and neck skin care preparations	10*	1	-0 1 1*	_
Body and hand skin care preparations	12	4	$\leq 0.1 - 1^{-1}$	_
Moisturizers	26	9	$\leq 0.1 - 1$	_
Night skin care preparations	20	1	$\leq 0.1 - 1$	_
Paste masks/mud packs	3	1	$\leq 0.1 - 1$	_
Skin fresheners	1	_	>0.1-1	_
Skin care preparations (other)	8	2	$\leq 0.1 - 1$	_
Suntan gels, creams, and liquids		1		—
Total uses/ranges for Polyamino Sugar Condensate	111	25	\leq 0.1–1	—

*This category was combined when the original safety assessment was performed and is now two separate categories.

Food and Drug Administration (FDA). 2001. Frequency of use of cosmetic ingredients. *FDA database*. Washington, DC: FDA.

Peterson, R. V., et al. 1986. Athymic nude mouse grafted with human skin as a model for evaluating the safety and effectiveness of radiolabeled cosmetic ingredients. J. Soc. Cosmet. Chem. 37:249–265.

Wenninger, J. A., R. C. Canterbery, and G. N. McEwen. Jr., eds. 2000. International Cosmetic Ingredient Dictionary and Handbook, 8th ed., 1114–1115. Washington, DC: CTFA.

POLYBUTENE

In 1982, CIR issued a Final Report that Polybutenes are safe as presently used in cosmetics (Elder 1982). One new inhalation toxicity study was reported since then (Skyberg et al. 1990). This new study, along with the updated information below regarding types and concentrations of use, were considered by the CIR Expert Panel. The Panel determined not to reopen this safety assessment.

In 1976 Polybutene was reported to be used in 84 cosmetic preparations, with the largest single use occurring in lipstick at concentrations of >1% to >50%. As reported to the FDA (FDA 2001), Polybutene is currently used in 253 products, with lipstick again the largest category and highest concentration, according to an industry survey (CTFA 2001). Table 20 presents the available use information.

The Panel noted that use concentration has increased overall and may currently be as high as 92% in some lipstick products, but that the available data demonstrate that this ingredient is not absorbed in the skin or the gut. The levels of Polybutene that are toxic via an inhalation route are not reached in cosmetics, and there are no aerosolized cosmetic products that contain Polybutene.

REFERENCES

Cosmetic, Toiletry, and Fragrance Association (CTFA). 2002. Product use concentration information for Polybutene. Unpublished data submitted by CTFA.²¹

²¹Available from the Director, Cosmetic Ingredient Review, 1101 17th Street NW, Suite 310, Washington, DC 20036.